

Dugway Proving Ground



DPG



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Report Documentation Page				Form Approved OMB No. 0704-0188	
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1. REPORT DATE 01 OCT 2005		2. REPORT TYPE N/A		3. DATES COVERED -	
4. TITLE AND SUBTITLE Inactivation of Bacillus anthracis Spores in Drinking Water by Mixed Oxidant Solution				5a. CONTRACT NUMBER	
				5b. GRANT NUMBER	
				5c. PROGRAM ELEMENT NUMBER	
6. AUTHOR(S)				5d. PROJECT NUMBER	
				5e. TASK NUMBER	
				5f. WORK UNIT NUMBER	
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) US Army Dugway Proving Ground				8. PERFORMING ORGANIZATION REPORT NUMBER	
9. SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS(ES)				10. SPONSOR/MONITOR'S ACRONYM(S)	
				11. SPONSOR/MONITOR'S REPORT NUMBER(S)	
12. DISTRIBUTION/AVAILABILITY STATEMENT Approved for public release, distribution unlimited					
13. SUPPLEMENTARY NOTES See also ADM001851, Proceedings of the 2003 Joint Service Scientific Conference on Chemical & Biological Defense Research, 17-20 November 2003. , The original document contains color images.					
14. ABSTRACT					
15. SUBJECT TERMS 23					
16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF ABSTRACT UU	18. NUMBER OF PAGES 43	19a. NAME OF RESPONSIBLE PERSON
a. REPORT unclassified	b. ABSTRACT unclassified	c. THIS PAGE unclassified			



Inactivation of *Bacillus anthracis* Spores in Drinking Water by Mixed Oxidant Solution

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Background

- Soldiers in some tactical scenarios do not have access to quartermaster potable water supplies.
- When calcium hypochlorite is used as a water disinfectant, a high free available chlorine concentration (FAC) is required to kill *B. anthracis* spores.
- Hypochlorite is toxic at levels around 190 mg/kg body weight.



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Background, cont.

- Water treated with calcium hypochlorite at a level to kill anthrax spores is neither potable nor palatable.



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Chemistry

- FAC is the concentration of $\text{Cl}_2 + \text{HOCl} + \text{OCl}^-$ (expressed as Cl_2) and is the most chemically reactive and biocidal form of chlorine commonly used for disinfection
- Hypochlorous acid (HOCl) is the predominant form of FAC at $\text{pH} < 7.2$.
- Optimum biocidal effectiveness of FAC occurs at pH 6.0 to 7.5.



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Chemistry, cont.

- At high pH HOCl dissociates primarily to hypochlorite ion ($\text{OCl}^- + \text{H}^+$).
- OCl^- is about 1/100 as effective a biocide as is HOCl, but still effective.
- HOCl at $\text{pH} < 4$ forms chlorine gas (Cl_2).
- Combined chlorines are formed from the reaction of FAC with nitrogenous compounds such as ammonia and urea.



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Chemistry, cont.

- Reactions with ammonia or amino group-containing organic compounds produce chloramines and organic chloramines.
- Chloramines are formed in a step-wise manner as FAC is added resulting in
 - Monochloramine (NH_2Cl)
 - Dichloramine (NHCl_2)
 - Trichloramine (NCl_3)



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Chemistry, cont.

- Chloramines are weak oxidizing agents and biocides relative to both HOCl and OCl^- .
- One source suggests that “monochloramine may only have 5% of the disinfecting power of FAC on a time/concentration basis.”



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Mixed Oxidants

- Produced by the electrolysis of sodium chloride solution
- Complete characterization of the molecular species present has not been possible.
- Characteristics of mixed oxidant activity suggest species are present in addition to HOCl and OCl⁻ (ie. FAC)



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Mixed Oxidant Advantages

- Rapid disinfection rates
- Removal of biofilms from surfaces of water distribution systems and treatment works
- More stable chlorine residual



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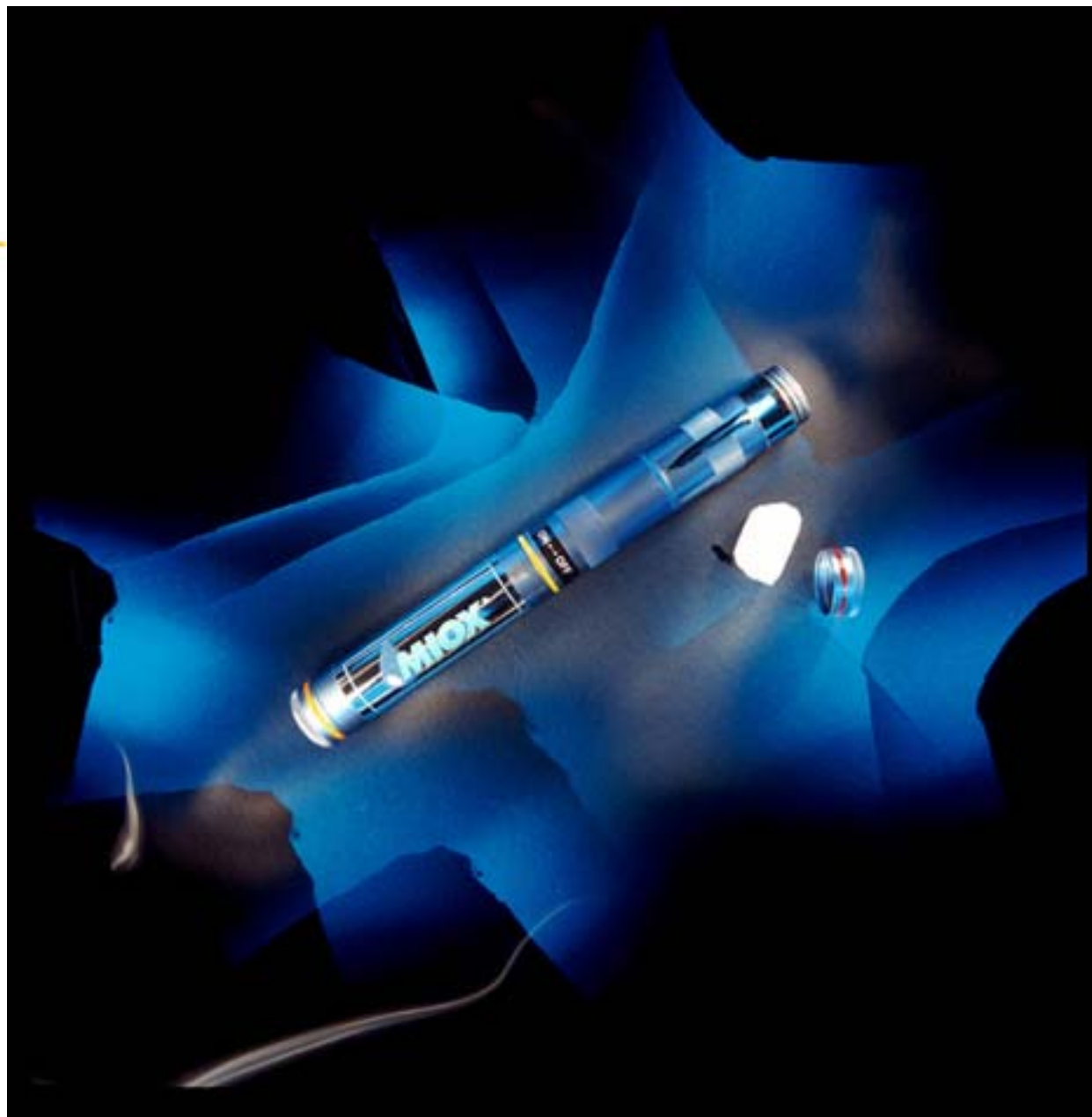
Disinfection Pen

- MIOX Corporation
- Los Alamos Technical Associates funded by DARPA to investigate technology for single soldier use
- Life Sciences Division tasked to provide independent evaluation using simulants and vaccine strains.



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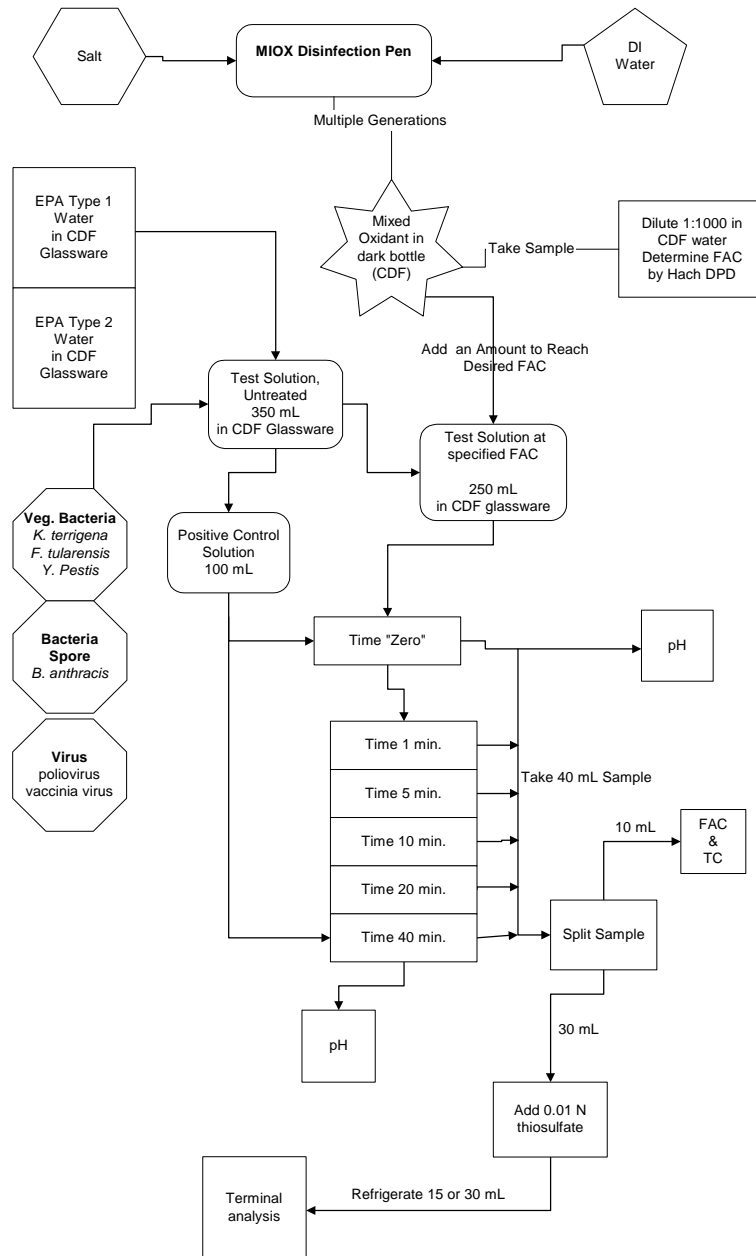
Initial Dugway Studies

- EPA Type 1 Water
 - Free of any chlorine or other disinfectant residue
 - pH 6.5 to 8.5
 - Total organic carbon 0.1 - 5.0 mg/L
 - Turbidity 0.1 to 5 NTU
 - Temperature $20 \pm 5^{\circ}\text{C}$
 - Total dissolved solids (sea salt) 50 - 100 mg/L



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Initial Studies, cont.

- *Bacillus anthracis* strain Sterne
 - spores
 - washed
- Looked for a 6 log reduction
- Challenge dose of 1E7 cfu/mL
- Details of test published as WDTC Document WDTC-IR-01-047



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Initial Studies, cont.

- Result: The 7 log cfu/mL challenge dose was killed in 40 minutes when treated with MOS at a FAC of 160 mg/L
- Analysis: It would take a soldier 40 MOS generation cycles of the pen to produce the required does in a 1-liter canteen.
- Conclusion: Impractical



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Follow-on Studies

- *B. anthracis* vegetative cells known to be sensitive to chlorine inactivation
- Pre-treat water with a germinant solution
- Phil Hanna (Univ. of Michigan) suggested
 - 10 mM L-alanine
 - 1 mM serine
 - 1 mM inosine

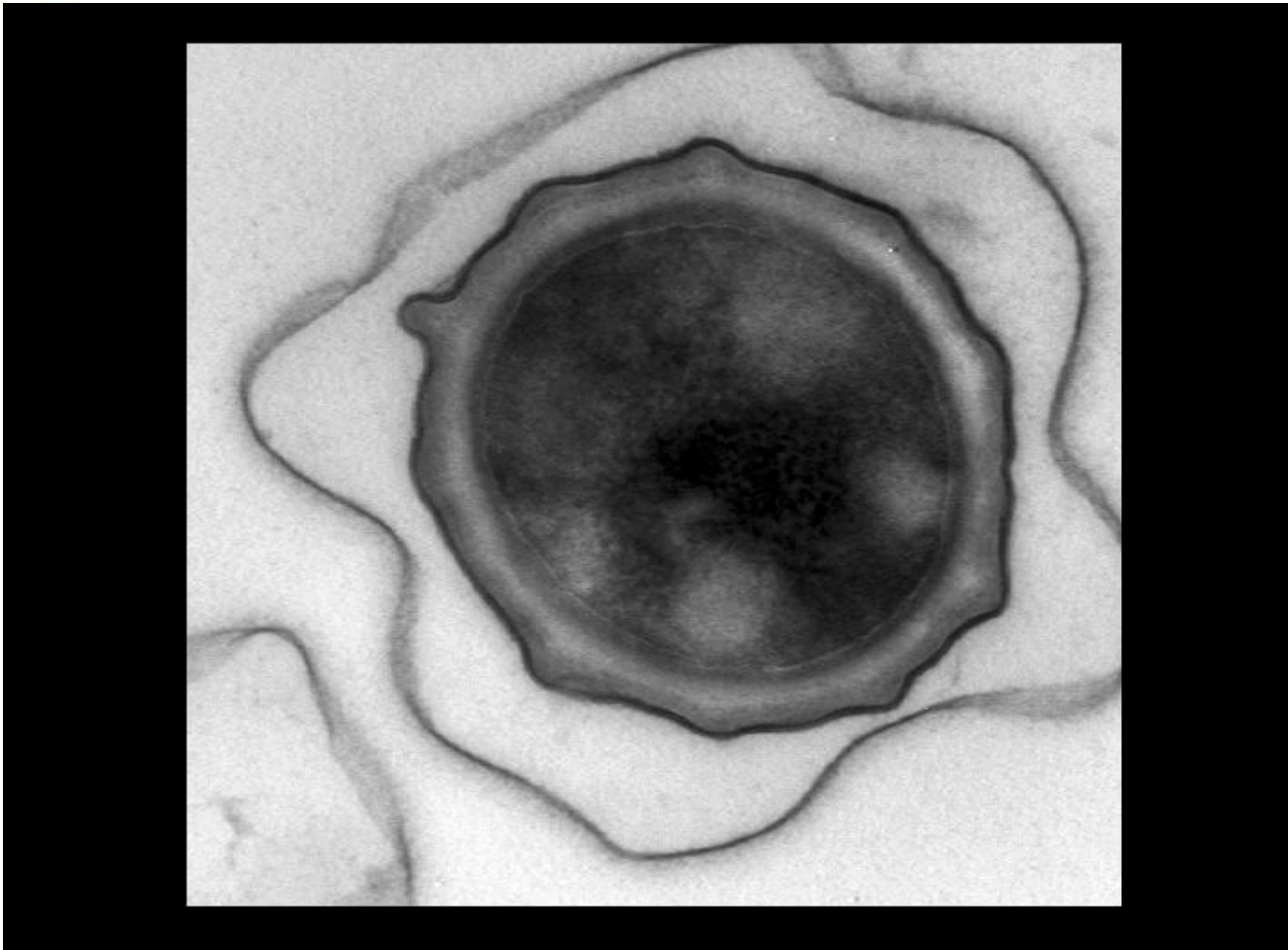


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Anthrax Spore Ultrastructure

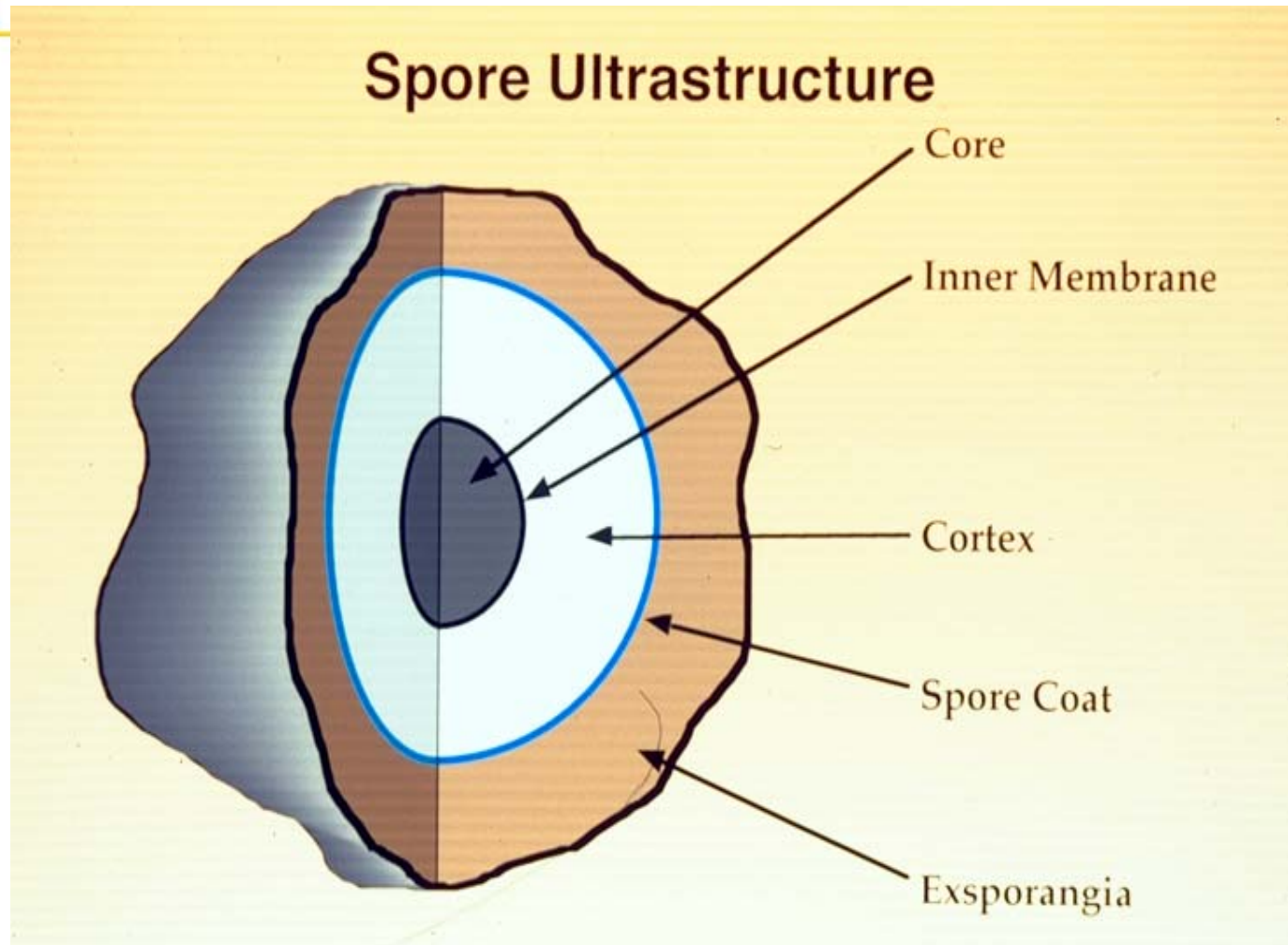


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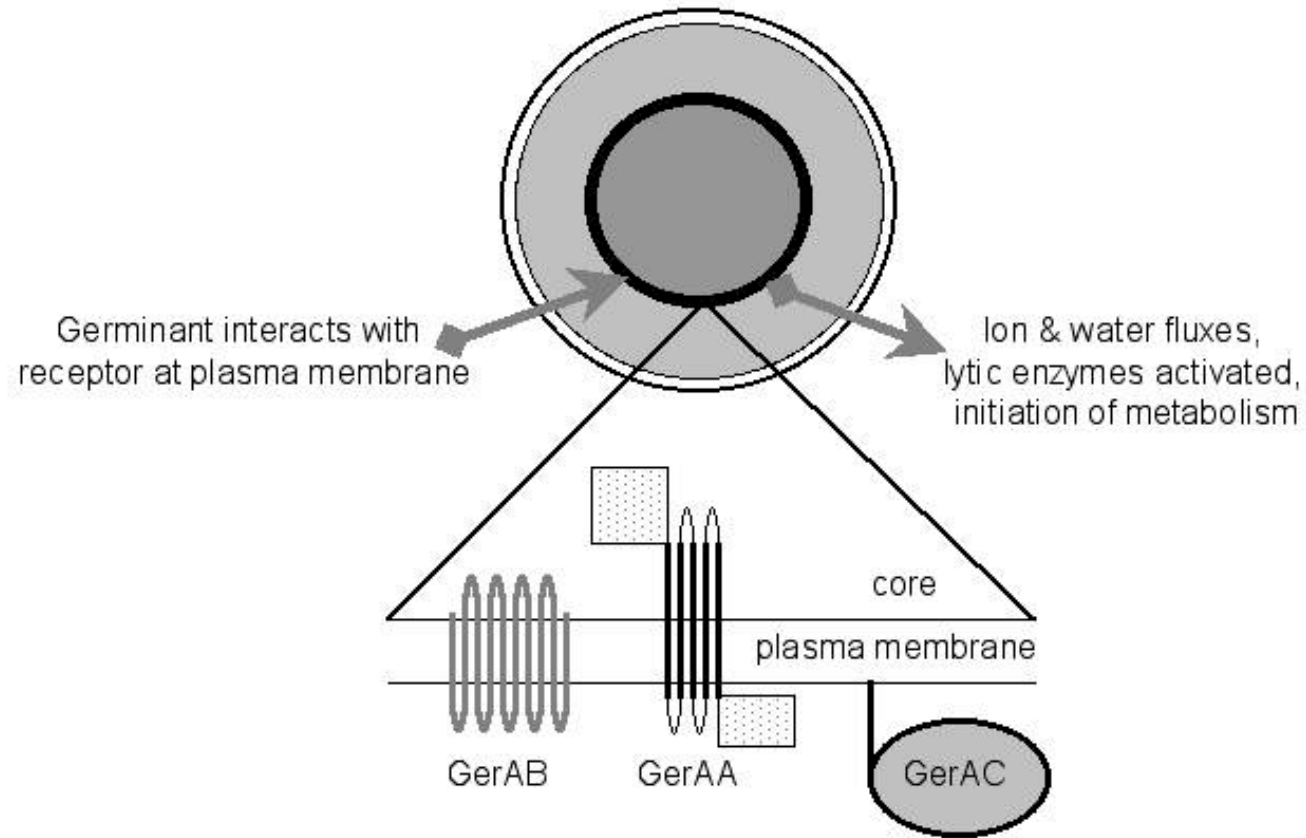


Anthrax Spore Ultrastructure



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Germinant Method A

- Used sterile tap water (no chlorine demand)
- Add Sterne spores to concentration of 1E6 cfu/mL
- Add germinant, incubate 30 minutes
- Add MOS to calculated FAC of 160 mg/L
- Plate samples at 1, 5, 10, 15, 25, 35, 45, & 60 minutes; incubate overnight



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Germinant + MOS Results

- Sterne spores would germinate using concentrations 1/10 of those recommended:
 - 1 mM L-Alanine
 - 0.1 mM Serine
 - 0.1 mM Inosine
- Our measure of germination being sensitivity to MOS.
- Germination results consistent with those previously published



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Method A Results, cont.

- Chlorine demand, germinant ~ 155 mg/L over 25 minutes
- Spores killed in 25 minutes after adding MOS at a final FAC decreasing to ~5.
- Significant increase in killing power of MOS on germinated spores
- Not unexpected



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Germinant Method B

- Same as Method A except that germinant and MOS were added to the spore suspension within less than 10 seconds of each other



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Method B Results

- Same as from Method A
- Spores killed in 20-30 minutes without a significant (<10 sec) incubation in germinant.
- **Unexpected result**
- No biochemical explanation evident
- Obvious that the majority of the FAC was being combined with germinants to form organic chloramines almost instantaneously.



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Organic Studies B

- The preceding results suggested another set of experiments.
- Looking at our results not as a function of germination but as a function of organic chloramine species and the N content of the germinants.



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Organic Chloramine Studies, cont.

- Theoretical framework from water treatment chemistry
 - A FAC ratio of 5 x N mass concentration (ie. 5 mg FAC/mg N) + 25% should generate organic monochloramine with negligible amounts of organic di- and trichloramine
 - The 25% additional FAC satisfied non-N oxidant demand, determined separately.



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Organic Chloramine Studies, cont.

- A FAC ratio of $7 \times \text{N}$ mass concentration + 25% should generate a combination of organic chloramines, NOT primarily monochloramine
- A FAC ratio of $7.6 \times \text{N} + 25\%$ should drive the reaction to breakpoint where the N is released from the chloramines as N_2 and FAC is reduced to Cl^- (ie. both FAC and chloramines reduced in concentration in the water)



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Organic Chloramine Experiments

- Prepare germinant solution such that $N = 21$ mg/L in $1E6$ cfu/mL Sterne spore suspension
- Dose with $5 \text{ FAC} \times N$ or $7 \text{ FAC} \times N$
- React for 30 minutes at room temperature
- Dose with $7.6 \times N$ to cause breakpoint
- Plates samples at 1, 5, 15, 25, 40, 60, & 90 minutes (no thiosulfate added)



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L-Ala, Ser, & Inosine: 5 x N

Time (min)	FAC	TC	cfu/mL
1	40	79	5.9E3
5	21	51	1E2
15	14	30	0



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L-Ala, Ser, & Inosine: 7 x N

Time (min)	FAC	TC	cfu/mL
1	17	32	9E1
5	11	23	3E1
15	7	17	0



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Organic Chloramine Conclusions

- Both the 7 x N and 5 x N doses against the complete germinant kill the spores within 15 minutes.
- The 5 x N dose has a flatter reaction curve.
- No clear difference between a solution presumably consisting of monochloramines versus a chloramine cocktail.



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Inorganic Chloramine Study

- Substitute ammonium chloride as the nitrogen source because chloramine generation from this molecule simpler in concept.
- Believed this experiment would be a negative control: assumed organic germinants possess some inherent quality required to explain previous results



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Inorganic Chloramine Experiment

- Ammonium chloride (NH_4Cl) substituted for germinant solution at same nitrogen concentration (21 mg/L)
 - Preston & Douthit (J Gen Microbiol, 1984) described its co-germinant effects on *B. cereus*.
- A sample was also taken after 15 min treatment and before the second or “breakpoint” 7.6 x N dose was added.



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NH_4Cl : 5 x N

Time (min)	FAC	TC	cfu/mL
1	9	47	0
5	8	22	0
15	7	17	0



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NH_4Cl : 7 x N

Time (min)	FAC	TC	cfu/mL
1	5	10	0
5	5	9	0
15	4.5	7.3	0



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15 Minute Samples

- $5 \times N = < 10 \text{ cfu/mL}$
- $7 \times N = 0 \text{ cfu/mL}$



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Inorganic Chloramine Conclusions

- A solution of chloramines derived from an inorganic source is capable of inactivating Sterne spores within 15 minutes.
- The effect was evident at 4 mg/L $\text{NH}_4\text{-N}$ (down from 21 mg) but not at 2 mg/L suggesting a very pronounced activation threshold.
- Monochloramines are probably the most active species.



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Inorganic Conclusions, cont.

- At 4 mg/L $\text{NH}_4\text{-N}$, the MOS dose required (38 mg/L FAC) is practical using the latest version of the Disinfection Pen.
- Preston and Douthit (1983, 1988) conducted experiments with *B. cereus* in which the germinant activity of NH_4Cl was studied in the presence of Ala and other known co-germinants.
- Our results suggest that Sterne spores are sensitive to NH_4Cl alone, probably at some point downstream from the recognition points for organic co-germinants



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Inorganic Conclusions, cont.

- This effect has been confirmed in preliminary experiments at Phil Hanna's lab in the absence of MOS.



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Acknowledgements

- Tim Dvorak (Dugway) for technical support in the initial Disinfection Pen experiments
- Phil Hanna (U. Mich.) for advice, encouragement and use of some of his figures.
- MIOX Corporation for provision of Disinfection Pens and figures.
- DARPA for project funding.



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